

10607 台北市大安區基隆路四段 43 號

No. 43, Keelung Road, Section 4, Taipei, Taiwan

National Taiwan University of Science and Technology

2020 Summer Program

CHEM 101 Introduction to Chemistry with Lab

Course Outline

Term: July 06-August 07,2019

Class Hours: 10:00-11:50 (Monday through Friday)

Course Code: CHEM 101

Instructor: Dr. Rodriguez

Home Institution: American University

Office Hours: 1:00PM to 1:30PM (Monday through Friday) and by appointment

Email: srodrigu@american.edu

Credit: 4

Class Hours: According to the regulations of Minister of Education, R.O.C, 18 class hours could be counted as 1 academic credit in all universities in Taiwan. This course will have 72 class hours, including 40 lecture hours, professor 10 office hours, 10-hour TA discussion sessions, 2-hour review sessions, 10 laboratory hours.

Course Description:

Introduction to Chemistry is a general introduction to chemistry, which is the study of matter and the changes it undergoes. In this course, you will learn the language of chemistry by using chemical symbols, formulas and equations, and use this language to understand the composition of materials, their structures and properties and related energy conversions. We will also cover practical applications of chemistry to problems involving environmental pollution, energy sources and human health. To keep up with the changing times, concepts of sustainability and green chemistry will also be addressed. Overall, skills gained in this course can be exceptionally useful in many aspects of your life.



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Course Objectives:

After successfully completing this class, students will be able to:

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- 1. Recognize that all matter is comprised of atoms, which have constant proportionality within molecules.
- 2. Demonstrate a basic understanding of chemical measurement.
- 3. Present a qualitative understanding of atomic and molecular structure, and the properties that arise from those structures.
- 4. Identify properties associated with chemical change.
- 5. Utilize the skills and methods learned in the course to make a logical argument or connection to a broader problem.
- 6. Explain the importance of chemistry in everyday life.

Required Textbook and Materials:

- Required Textbook: The Molecular World by OpenStax College (Adapted by Michele Lansigan) © 2016 by Adapted at American University (downloadable FREE OF CHARGE at http://chem100textbook.openbooks.wpengine.com/)
- 2. Scientific Calculator

Grading & Evaluation:

| Criteria | Points |
|------------|--------|
| Homework | 20% |
| Midterm | 20% |
| Lab | 25% |
| Final Exam | 35% |
| FOTAL | 100% |



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• Lectures will cover course material and the homework for the week will be related to the material covered in those lectures. You will be responsible for completing the assigned readings and homework prior to each lecture.

• Weekly homework will be given based on the material covered in class and practice problems in the textbook. It will be approximately 10 problems per week.

• Exams will cover the material discussed in class and homework problems. The final exam will be comprehensive, covering material from the entire 5 weeks. ONLY scientific calculator will be allowed for exams.

• The lab is designed to develop the ability to integrate concepts and analyze scientific data.

• Since your ideas and perspectives are a valuable part of the course dialogue, students are expected to attend class and remain in attendance for a full class period. Partial attendance means coming late, leaving early, or leaving and returning for a prolonged period of time within one session. If you are 15 minutes late, you are considered absent

| | Date | Торіс |
|--------|------------|---|
| Week 1 | Lecture 1 | Course Introduction / Chapter 1: Essential Ideas in Chemistry |
| | Lecture 2 | Chapter 2: Atomic Theory and Structure |
| | Lab | Literature Review |
| Week 2 | Lecture 3 | Chapter 3: Electronic Structure |
| | Lecture 4 | Chapter 4: Chemical Bonds |
| | Lecture 5 | Chapter 5: Composition and Solutions |
| | Lab | Experiment Proposal |
| Week 3 | Lecture 6 | Chapter 6: Stoichiometry of Chemical Reactions |
| | Lecture 7 | Practice Problems |
| | Lecture 8 | Midterm |
| | Lab | Collecting Data |
| Week 4 | Lecture 9 | Midterm |
| | Lecture 10 | Chapter 7: Acid and Bases |
| | Lecture 11 | Chapter 8: Gases |
| | Lab | Data Analysis |
| Week 5 | Lecture 12 | Chapter 8: Electrochemistry |

<u>Course Schedule (tentative):</u>



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| Lecture 13 | General Review |
|------------|---------------------------|
| Lecture 14 | Final Exam |
| Lab | Writing Scientific Report |

